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Bioactivity-Guided Synthesis Accelerates the Discovery of 3-(Iso)quinolinyl-4-chromenones as Potent Fungicide Candidates

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ABSTRACT: Fungal infections could cause tremendous decreases in crop yield and quality. Natural products, including flavonoids and (iso)quinolines, have always been an important source for lead discovery in medicinal and agricultural chemistry. To promote the discovery and development of new fungicides, a series of 3-(iso)quinolinyl-4-chromenone derivatives was designed and synthesized by the active substructure splicing principle and evaluated for their antifungal activities. The lead optimization was guided by bioactivity. The bioassay data revealed that the 3-quinolinyl-4-chromenone 13 showed significant *in vitro* activities against *S. sclerotiorum*, *V. mali*, and *B. cinerea* with EC₅₀ values of 3.65, 2.61, and 2.32 mg/L, respectively. The 3-isoquinolinyl-4-chromenone 25 exhibited excellent *in vitro* activity against *S. sclerotiorum* with an EC₅₀ value of 1.94 mg/L, close to that of commercial fungicide chlorothalonil (EC₅₀ = 1.57 mg/L) but lower than that of boscalid (EC₅₀ = 0.67 mg/L). For *V. mali* and *B. cinerea*, 3-isoquinolinyl-4-chromenone 25 (EC₅₀ = 1.56, 1.54 mg/L) showed significantly higher activities than chlorothalonil (EC₅₀ = 11.24, 2.92 mg/L). In addition, *in vivo* experiments proved that compounds 13 and 25 have excellent protective fungicidal activities with inhibitory rates of 88.24 and 94.12%, respectively, against *B. cinerea* at 50 mg/L, while the positive controls chlorothalonil and boscalid showed inhibitory rates of 76.47 and 97.06%, respectively. Physiological and biochemical studies showed that the primary action of mechanism of compounds 13 and 25 on *S. sclerotiorum* and *B. cinerea* may involve changing mycelial morphology and increasing cell membrane permeability. In addition, compound 13 may also affect the respiratory metabolism of *B. cinerea*. This study revealed that compounds 13 and 25 could be promising candidates for the development of novel fungicides in crop protection.

KEYWORDS: natural product, flavonoid, (iso)quinoline, fungicidal activity, structure-activity relationship

INTRODUCTION

Fungal infection could cause tremendous decreases in crop yield and quality, which threatens worldwide food security.^{1,2} Treating crops with fungicides has been one of the most effective ways against plant diseases. Nevertheless, widespread use and misuse of chemical fungicides have caused more and more resistance in the fungi.^{3,4} Meanwhile, green fungicides with high efficiency and low toxicity are needed to reduce the negative impact on the environment. Using natural products as lead compounds to develop new fungicides is one of effective solutions to these problems.^{5,6} Natural products with novel structures, as a source for new pesticides, not only can provide unique modes of action⁷ but also have good environmental compatibility, which is important for development of green fungicides.⁸

Natural flavonoids have been found in many plants, such as tea,⁹ ginkgo,¹⁰ *Hippophae rhamnoides*,¹¹ *Flos Sophorae Immaturus*, etc.^{9,12} Furthermore, flavonoids exhibit a variety of biological activities including antitumor,^{13,14} anti-inflammatory,¹⁵ antioxidation,¹⁶ anticarcinogenic,¹⁷ anti-HIV,¹⁸ insecticidal,¹⁹ and fungicidal effects (Figure 1).^{20,21} Containing a valuable structural core, flavonoids have always been used as lead compounds in the development of medicines and pesticides.²²

Meanwhile, (iso)quinolines, one of the largest classes of alkaloids, take a great proportion of the active natural products.²³ The first quinoline was isolated from the bark of

the Cinchona tree in 1820.²⁴ To date, there are more than 150 (iso)quinoline alkaloids isolated,²⁵ which show a wide range of biological effects, such as antitumor,²⁶ antimalarial,²⁷ antibacterial,²⁸ and antifungal activities.²⁹ Also, (iso)quinolines have been identified as a pharmacophore in many complex natural products, which exhibit highly antifungal activities. They have been considered as potential active groups and a functional scaffold reasonably in agrochemical discovery. For instance, the quinoline skeleton has been imbedded in the marketed fungicides tebufloquin³⁰ and quinofumelin (Figure 1).^{31,32}

Recently, We have developed a general approach for the synthesis of 3-(iso)quinolinyl 4-chromenones,³³ which displayed obvious antifungal activities during our preliminary biological screening. Motived by this, in the present study, we attempted to further investigate flavonoid-based antifungal agents. Quercetin was taken as the lead compound for systematic structural optimization. Attracted by the known fungicidal activities of (iso)quinoline, we expected that the antifungal effects of the lead compound could be further

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Figure 1. Chemical structures of flavonoids (A) and (iso)quinolines (B) with biological activities.



Figure 2. Design of the target compounds.

improved by introducing (iso)quinoline to the flavonoid skeleton through an active substructure splicing strategy. With this in mind, series of novel 3-(iso)quinolinyl-4chromenone derivatives were designed and synthesized (Figure 2). The fungicidal activities of these novel compounds were evaluated to validate the idea, and the structure–activity relationships (SARs) were analyzed. The results of biological activity assays indicate that many of target compounds exhibit excellent fungicidal activities, which initially verified the rationality of our design strategy. To explore the preliminary mechanisms of the action of 3-(iso)quinolinyl-4-chromenone derivatives, the morphological changes of the mycelia cell wall and the antifungal effects on the permeability of the mycelia cell membrane and respiratory metabolism of fungi were examined.

MATERIALS AND METHODS

Instruments and Chemicals. All chemicals (reagent grade) were purchased from commercial sources (Energy, Shanghai, China). All the ¹H and ¹³C NMR spectra were measured on a Bruker AV-400, AV-500, or AV-600 spectrometer with $CDCl_3$, CD_2Cl_2 , or DMSO- d_6 as the solvent and tetramethylsilane as the internal standard. Chemical shifts were reported in ppm (δ). Melting points (mp) were recorded

on a WRS-1B melting point apparatus (Jingsong, Shanghai, China) and were measured without correction. High-resolution mass spectrometry (HRMS) was recorded using an electrospray ionization (ESI) technique. Thin-layer chromatography (TLC) was performed on a silica gel 60 GF254 plate. The silica gel (size of 200–300 mesh) used for the column chromatography was purchased from Qingdao Haiyang Chemistry Plant (China). Microscopic morphology of fungal hyphae was observed by a scanning electron microscope (Hitachi, S-3400 N, Tokyo, Japan). The conductivity was measured by a CONS10 Eutech/Oakton conductometer (OAKTON Instruments, Waltham). Dissolved oxygen was measured by a JPB-607A dissolved oxygen meter (INESA Scientific Instrument Co., Shanghai, China).

Fungi. Colletotrichum orbiculare (C. orbiculare), Fusarium oxysporum (F. oxysporum), Sclerotinia sclerotiorum (S. sclerotiorum), Physalospora piricola (P. piricola), Valsa mali (V. mali), Altenaria alternariae (A. alternariae), and Botrytis cinerea (B. cinerea) were provided by the College of Plant Protection, Northwest A&F University.

Synthesis. General Procedure for the Preparation of Intermediates 1a-h. n-Butyllithium (2.0 mL, 5.0 mmol, 2.5 M in hexane) was added dropwise to the solution of alkyne (5 mmol) in dry THF (30 mL) at -78 °C. The solution was stirred at this temperature under argon for 1 h. After that, salicylaldehyde (2.27 mmol) was added using a syringe at the same temperature; the solution was stirred for 2 h at -78 °C. Then, the reaction was quenched by the

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Figure 3. (A) General synthesis of 1a-h. Reagents and conditions: (a) *n*-BuLi, THF, -78 °C; MnO₂, acetone. (B) General synthesis of 3–22. Reagents and conditions: (b) quinoline *N*-oxide, DMF, 140 °C, HCl. (C) General synthesis of 23–33. Reagents and conditions: (c) isoquinoline *N*-oxide, DMF, 140 °C, HCl.

addition of aqueous saturated NH₄Cl (20 mL). The phases were separated and the aqueous phase was extracted with EtOAc (3×10 mL). The combined organic layers were washed with brine (30 mL) and dried over Na₂SO₄. The resulting liquid was concentrated under a reduced pressure. Then, the crude material was dissolved in 30 mL of acetone, and activated MnO₂ (15.0 g, 172.5 mmol) was added to the mixture. Then, the reaction mixture was stirred for 12 h at room temperature, filtered through a pad of celite, concentrated under a reduced pressure, and purified by flash chromatography on silica gel (petroleum ether:ethyl acetate = 60:1) to afford the product.

General Procedure for the Preparation of Compounds 3–22. Intermediate 1 (1 mmol) and quinoline *N*-oxide (1.5 mmol) are dissolved in 1.5 mL of DMF in a 20 mL sealed tube; then, 25 μ L of 12 M HCl solution was added into the mixture quickly. The reaction mixture was stirred at 140 °C in a dry block heater for 2 h. After that, the solvent was removed, and the crude product was purified via column chromatography on silica gel (Gradient elution: petrol ether:ethyl acetate = 15:1 to dichloromethane:methanol = 100:1) to give the corresponding product.

General Procedure for the Preparation of Compounds 23–33. Intermediate 1 (1 mmol) and isoquinoline N-oxide (1.5 mmol) are dissolved in 1.5 mL of DMF in a 20 mL sealed tube; then, 25 μ L of 12 M HCl solution was added into the mixture quickly. The reaction mixture was stirred at 140 °C in a dry block heater for 2 h. After that, the solvent was removed, and the crude product was purified via column chromatography on silica gel (Gradient elution: petrol ether:ethyl acetate = 15:1 to dichloromethane:methanol = 100:1) to give the corresponding product.

Biological Assay. In Vitro Fungicidal Activities. All synthesized compounds were screened for their *in vitro* antifungal activities against *C. orbiculare, F. oxysporum, S. sclerotiorum, P. piricola, V. mali, A. alternariae,* and *B. cinerea* at the concentration of 50 mg/L for the preliminary screening according to a mycelia growth inhibition method, with quercetin, chlorothalonil, tebuconazole, procymidone, and boscalid as positive controls.

In Vivo Fungicidal Activities against B. cinerea. The Cucumis sativus Linn. leaves of rape were collected from the Key Laboratory of Botanical Pesticide R&D in Northwest A&F University. For the protective activity assay, healthy leaves of Cucumis sativus Linn. were sprayed with the target compounds (50 mg/L), respectively, and then cultivated at 25 °C for 24 h before inoculation with B. cinerea. Chlorothalonil and boscalid were used as the positive controls.

Effects of Compounds 13 and 25 on Mycelial Morphology of S. sclerotiorum and B. cinerea. S. sclerotiorum and B. cinerea were cultured in a 90 mm culture dish. When the mycelia grew to 70 mm in diameter, fresh fungus dishes (5 mm in diameter) were made from the edge of the colonies. The mycelia were inoculated on potato dextrose agar (PDA) medium plates containing no compound (negative control) and compounds 13 and 25 with a concentration of 5 mg/L,

respectively. Then, the mycelia were cultured at 25 $^{\circ}$ C for 2 days. The mycelia cell wall structure of *S. sclerotiorum* at the top of each treated colonies were selected and observed under an S-3400 N scanning electron microscope (SEM) (Hitachi, Ltd., Tokyo, Japan).

Effects of Compounds 13 and 25 on Cell Membrane Permeability of S. sclerotiorum, V. mali, and B. cinerea. The tested strains were cultured on a PDA plate at 25 °C for 72 h. Then, the strains were placed in a potato dextrose broth (PDB) culture medium and cultured by shaking (25 °C, 120 rpm) for 72 h. The mycelia were filtered and washed with distilled water and put into a centrifugal tube, and then 10 mL of compounds 13 and 25 (25 mg/L) and distilled water were added into the centrifugal tube, respectively. Finally, the mycelia were oscillated (120 rpm) in a water bath at a constant temperature of 28 °C at different times. The conductivity was measured by a CON510 Eutech/Oakton conductometer (OAKTON Instruments, Waltham). The negative control was mycelia with distilled water. The conductivity of compounds 13 and 25 were determined at 0, 30, 60, 120, 180, 240, and 300 min and finally boiled (dead treatment) to determine the conductivity. The relative permeability was calculated for each measurement, and then the permeability of cell membranes was compared according to the conductivity. Each experiment was run in triplicate.

Effects of Compounds **13** and **25** on the Respiratory Metabolism of B. cinerea. The respiratory oxygen consumption rate of B. cinerea was measured by the oxygen electrode method.³⁴

A total of 500 mg of fresh mycelia were added into 10 mL of medium solution (9 mL of phosphate buffer solution with a concentration of 1 M, pH = 7.2 and 0.2 mL of glucose solution with a concentration of 2%). After stirring for 10 min, dissolved oxygen was measured by a JPB-607A dissolved oxygen meter (INESA Scientific Instrument Co., Shanghai, China).

Then, compounds **13** and **25**, and boscalid (20 mg/L) were added. Respiration rates of mycelia (O_2 , μ mol/g min) were calculated from the change of oxygen content in the medium. The inhibition rates of mycelia respiration (IR_r) were calculated according to the respiration rates of mycelia before and after adding test agents with the following equation:

$$IR_r(\%) = (R_0 - R_1)/R_0$$

where R_0 and R_1 are respiration rates of mycelia before and after adding tested agents.

RESULTS AND DISCUSSION

Chemistry Synthesis. The synthetic route for compounds 1a-h was described in Figure 3A. The target compounds can be divided into two series: (1) 3-quinolinyl-4-chromenones (Figure 3B); (2) 3-isoquinolinyl-4-chromenones (Figure 3C). The chemical structures of all prepared compounds were confirmed by ¹H NMR, ¹³C NMR, and HRMS.

Fungicidal Activities and SAR Discussion. A bioactivityguided synthesis was applied for the discovery of antifungal candidates. All compounds were evaluated against seven fungi including *C. orbiculare, F. oxysporum, S. sclerotiorum, P. piricola, V. mali, A. alternariae,* and *B. cinerea.* This biologically indicating method is handleable and intuitionistic. The results generated from this tactic are easy to interpret with a significant value (Figure 4).

Initially, 3-quinolinyl-4-chromenones 3-14 were synthesized and evaluated for their antifungal activities. As can be seen from Table 1, most of the synthesized compounds exhibited fair to good antifungal activities against *S. sclerotiorum*, *V. mali*, and *B. cinerea*. Among them, compounds 8 ($R_2 = 5$ -OCH₃) and 13 ($R_2 = 6$ -OCH₃) displayed over 96% inhibitory activity, which indicated that the methoxy group at the quinoline moiety may be crucial for their significant *in vitro* antifungal bioactivity. The higher biological activity of the methoxy group may be related to its electron-donating effect.³⁵



Figure 4. Bioactivity guided synthesis of compounds 3-22.

To confirm this, the compounds 15 ($R_2 = 7$ -OCH₃), 16 ($R_2 =$ 8-OCH₃), and 17 $(R_2 = 3-OCH_3)$ were synthesized. Interestingly, they all show good biological activities against S. sclerotiorum, V.mali, and B.cinerea, which further proved that the methoxy group is a factor determining the biological activities of the target compounds. Comparing the biological activities of compounds 8, 13, 15, 16, and 17, we found that the compounds with a methoxy group at the C5/C6 position of quinoline exhibited relatively higher activities. However, because the cost of the starting material 6-methoxyquinoline (\$ 201/100 g, Energy) is much lower than that of 5methoxyquinoline (\$ 857/100 g, Energy), we chose compound 13 $(R_2 = 6 - OCH_3)$ as the lead compound for further optimization. Therefore, after R_2 is fixed as the 6-OCH₃ group, the influence of R_1 substituents at the 4-position of a phenyl ring on the antifungal activities was then investigated. The compounds 18, 19, 20, 21, and 22 were synthesized. The results show that compound 13 $(R_1 = H)$ exhibited the best antifungal activity compared with the corresponding compounds.

In order to investigate whether compounds 3-22 possess a broad spectrum of bioactivity, we selected four more fungi (C. orbiculare, F. oxysporum, P. piricola, and A. alternariae) to test their antifungal activities. For C. orbiculare, compound 17 (R₂ = 3-OCH₃, 80.81%) showed an excellent inhibitory rate (about 80%), which is better than that of other compounds. In addition, compounds 6 ($R_2 = 5$ -Cl, 72.22%) and 11 ($R_2 = 6$ -Cl, 76.77%) displayed good antifungal activities against C. orbiculare (inhibition rate > 70%). For F. oxysporum, although the inhibition rates of all compounds are below 70%, compound 17 with a methoxy group $(R_2 = 3-OCH_3)$ still showed the highest inhibition rate (67.62%), and compounds **15** ($R_2 = 7$ -OCH₃, 57.90%) and **16** ($R_2 = 8$ -OCH₃, 63.81%) showed moderate fungicidal activity compared with other compounds. For P.piricola, unfortunately, only compounds 3 $(R_2 = H, 52.22\%)$, 13 $(R_2 = 6$ -OCH₃, 50.00%), and 21 $(R_2 =$ 6-OCH₃, 55.21%) gave the inhibition rates over 50%.

To further optimize the leads, compounds 23–33 were designed by bioisosterism and evaluated for their antifungal effects against *S. sclerotiorum*, *V.mail*, and *B. cinerea* at 50 mg/L. As shown in Table 2, some of them displayed good fungicidal activity against specific fungi. Among them, compounds 23, 24, 25, 28, and 29 displayed good antifungal activities against *S. sclerotiorum* (inhibition rate > 80%) at 50 mg/L, which showed inhibition rates of 98.33, 98.06, 100, 84.44, and 98.33%, respectively. For the *V. mali*, compounds 23, 24, and 25 exhibited >90% inhibition rates. Notably, compounds 23–33 displayed good antifungal activities against *B. cinerea* (>80% inhibition rates). Of note, in the 3-isoquinolinyl-4-chromenone series, compound 25 showed the best antifungal activities against the seven fungi.

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Table 1. In Vitro Fungicidal Activities	(Inhibition Rate) of the Compounds 3–22 ($(50 \text{ mg/L})^{a,b,c,a,e,f,g,h,i}$
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comnda	Cab	E o ^c	s d	Dwe	Vmf	1 28	R _c ^h
compa	C.0	1.0	5.0	г.р	V.m	A.u [_]	D.t
3	50.56 ± 1.10	34.44 ± 2.86	50.56 ± 1.22	52.22 ± 1.01	65.64 ± 1.12	65.63 ± 0.53	90.01 ± 1.04
4	49.49 ± 0.88	30.48 ± 2.18	44.12 ± 2.25	<5	72.92 ± 1.38	59.90 ± 1.57	34.09 ± 2.23
5	61.11 ± 1.75	52.38 ± 0.83	93.14 ± 1.30	42.53 ± 3.77	93.75 ± 0.52	60.94 ± 1.56	37.12 ± 1.52
6	72.22 ± 0.87	38.10 ± 0.83	61.27 ± 0.98	35.06 ± 1.15	92.19 ± 0.90	59.38 ± 1.27	59.85 ± 0.76
7	51.01 ± 4.63	31.43 ± 1.43	47.06 ± 2.94	32.18 ± 1.52	88.02 ± 1.04	64.58 ± 1.80	37.12 ± 1.52
8	48.33 ± 1.65	26.11 ± 3.44	100	25.19 ± 3.92	100	50.00 ± 1.57	95.5 ± 0.55
9	47.98 ± 0.87	40.48 ± 1.65	81.25 ± 0.64	<5	75.00 ± 0.26	70.31 ± 0.90	71.67 ± 1.45
10	51.52 ± 0.81	56.19 ± 4.12	91.67 ± 0.28	<5	92.22 ± 0.13	91.67 ± 0.12	88.33 ± 1.22
11	76.77 ± 1.51	50.95 ± 2.98	72.22 ± 0.33	31.48 ± 0.37	54.36 ± 2.25	31.77 ± 0.50	73.91 ± 2.13
12	35.86 ± 1.75	36.19 ± 0.83	42.08 ± 3.11	19.35 ± 4.52	53.33 ± 2.12	61.46 ± 1.21	71.11 ± 1.03
13	48.48 ± 1.52	33.33 ± 0.72	100	50.00 ± 2.63	97.92 ± 0.22	63.02 ± 0.98	96.97 ± 0.31
14	45.57 ± 3.2	15.31 ± 0.90	46.67 ± 0.59	30.30 ± 1.61	39.49 ± 0.51	27.02 ± 1.56	48.48 ± 0.62
15	66.16 ± 0.68	57.90 ± 1.32	100	48.85 ± 2.17	93.75 ± 1.10	85.42 ± 0.89	98.48 ± 0.25
16	59.60 ± 2.31	63.81 ± 0.83	75.00 ± 1.20	27.01 ± 3.59	91.67 ± 0.65	90.63 ± 1.57	100
17	80.81 ± 0.88	67.62 ± 0.82	84.17 ± 0.85	43.68 ± 2.51	88.54 ± 0.75	75.00 ± 0.91	91.67 ± 1.52
18	45.96 ± 2.31	11.09 ± 2.72	38.33 ± 3.52	7.29 ± 5.11	37.84 ± 2.53	41.33 ± 1.42	<5
19	36.36 ± 1.51	10.95 ± 0.83	22.00 ± 4.25	18.75 ± 2.53	39.19 ± 2.47	39.33 ± 2.58	26.34 ± 3.55
20	46.46 ± 0.89	28.10 ± 1.69	34.67 ± 3.10	28.65 ± 4.88	59.46 ± 1.28	52.00 ± 2.02	32.80 ± 2.38
21	51.52 ± 1.52	45.71 ± 2.47	75.00 ± 1.02	55.21 ± 2.62	90.99 ± 0.42	75.33 ± 1.15	80.65 ± 0.66
22	46.46 ± 0.72	19.05 ± 3.86	41.67 ± 3.05	24.48 ± 4.55	55.41 ± 1.37	48.00 ± 1.65	23.66 ± 4.21
quercetin	33.33 ± 2.15	<5	100	8.62 ± 0.99	73.96 ± 1.80	<5	9.09 ± 4.32
C^{i}	70.73 ± 1.45	76.19 ± 0.55	100	86.15 ± 0.64	71.67 ± 1.65	66.67 ± 2.08	91.6 ± 0.58
boscalid	97.62 ± 0.48	50.00 ± 0.78	100	66.67 ± 0.60	62.72 ± 0.52	44.59 ± 0.78	85.76 ± 0.61
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^aData are given as the mean of triplicate experiments. ^bC.o: Colletotrichum orbiculare. ^cF.o: Fusarium oxysporum. ^dS.s: Sclerotinia sclerotiorum. ^eP.p: Physalospora piricola. ^fV.m: Valsa mali. ^gA.a: Altenaria alternariae. ^hB.c: Botrytis cinerea. ⁱC: chlorothalonil.

Table 2. In Vitro Fungicidal Activities ((Inhibition Rate)	of the Compounds 23–3	3 $(50 \text{ mg/L})^{a,b,c,d,e,f,g,h,i}$
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compd ^a	C.o ^b	F.o ^c	S.c ^d	P.p ^e	V.m ^f	A.a ^g	B.c ^h
23	56.67 ± 2.14	62.22 ± 2.04	98.33 ± 0.13	57.04 ± 0.37	97.95 ± 0.58	78.13 ± 0.90	96.14 ± 0.30
24	58.33 ± 1.65	68.44 ± 1.03	98.06 ± 0.20	52.22 ± 0.98	95.38 ± 0.14	76.56 ± 0.90	97.67 ± 0.13
25	65.00 ± 1.32	69.33 ± 0.28	100	63.33 ± 0.63	100	79.85 ± 0.14	100
26	54.30 ± 0.69	32.78 ± 2.57	72.22 ± 1.41	67.41 ± 0.97	82.05 ± 0.95	73.44 ± 1.82	88.56 ± 0.22
27	52.22 ± 1.25	41.11 ± 1.10	50.00 ± 2.44	41.11 ± 2.30	86.67 ± 0.83	75.00 ± 1.57	85.55 ± 0.55
28	65.00 ± 1.05	62.78 ± 1.42	84.44 ± 1.58	72.22 ± 0.63	87.69 ± 1.32	62.50 ± 0.14	90.34 ± 0.30
29	60.00 ± 0.45	20.00 ± 3.23	98.33 ± 0.22	3.33 ± 0.63	38.97 ± 2.75	34.90 ± 0.50	81.23 ± 0.79
30	45.00 ± 2.61	16.67 ± 3.57	42.78 ± 3.11	54.44 ± 0.42	47.18 ± 2.35	46.35 ± 0.53	86.71 ± 1.85
31	37.78 ± 2.78	40.00 ± 1.65	75.56 ± 2.02	42.59 ± 0.74	85.13 ± 0.88	59.90 ± 0.57	88.21 ± 0.52
32	40.00 ± 2.65	38.33 ± 2.41	46.67 ± 3.44	69.63 ± 0.98	63.59 ± 1.22	67.71 ± 0.53	92.12 ± 0.79
33	53.33 ± 0.88	62.78 ± 1.58	30.56 ± 4.20	42.96 ± 0.37	49.74 ± 1.38	66.15 ± 0.53	88.86 ± 0.60
quercetin	33.33 ± 2.15	<5	100	8.62 ± 0.99	73.96 ± 1.80	<5	9.09 ± 4.32
C ⁱ	70.73 ± 1.45	76.19 ± 0.55	100	86.15 ± 0.64	71.67 ± 1.65	66.67 ± 2.08	91.6 ± 0.58
boscalid	97.62 ± 0.48	50.00 ± 0.78	100	66.67 ± 0.60	62.72 ± 0.52	44.59 ± 0.78	85.76 ± 0.61
^a Data ara aiwa	n as the mass of t		bC a Callatation	leven autoinut and CE a	. Europium annona	de a Calanatini	e al anationum e Du

"Data are given as the mean of triplicate experiments. "C.o: Colletotrichum orbiculare. "F.o: Fusarium oxysporum. "S.s: Sclerotinia sclerotiorum. "P.p: Physalospora piricola. ^fV.m: Valsa mali. ^gA.a: Altenaria alternariae. ^hB.c: Botrytis cinerea. ⁱC: chlorothalonil.

In order to further determine the fungicidal potency and probe the SARs of these novel 3-(iso)quinolinyl-4-chromenone derivatives, several compounds with superior fungicidal activities were selected for further study. Their EC₅₀ values against *S. sclerotiorum*, *V. mali*, and *B. cinerea* are shown in Table 3. In general, the compounds bearing methoxy groups showed better fungicidal activities than that with other substituents. For example, compound 8 ($R_2 = 5$ -OCH₃, EC₅₀ = 3.79 mg/L) was more active than compound 5 ($R_2 = 5$ -F, EC₅₀ = 4.60 mg/L) and compound 13 ($R_2 = 6$ -OCH₃, EC₅₀ = 3.65 mg/L) was more active than compound 10 ($R_2 = 6$ -F, EC₅₀ = 7.50 mg/L) against *S. sclerotiorum*. Compound 13 ($R_2 = 6$ -OCH₃, EC₅₀ = 2.61, 2.32 mg/L) was more active than compound 10 ($R_2 = 6$ -F, EC₅₀ = 3.47, 4.72 mg/L) against *V*. *mali* and *B. cinerea*, respectively. Meanwhile, the compounds with substitution at the 5/6-position of the quinoline ring showed better fungicidal activities than that with substitution at other positions. For instance, compounds 8 ($R_2 = 5$ -OCH₃, $EC_{50} = 3.79$, 2.37 mg/L) and 13 ($R_2 = 6$ -OCH₃, $EC_{50} = 3.65$, 2.32 mg/L) were more active than compounds 15 ($R_2 = 7$ -OCH₃, $EC_{50} = 5.27$, 2.47 mg/L), 16 ($R_2 = 8$ -OCH₃, $EC_{50} = 8.55$, 6.54 mg/L), and 17 ($R_2 = 3$ -OCH₃, $EC_{50} = 18.18$, 13.54 mg/L), against *S. sclerotiorum* and *B. cinerea*. For *V. mali*, compound 8 ($EC_{50} = 1.65$ mg/L) was more active than compounds 13 ($EC_{50} = 2.61$ mg/L), 17 ($EC_{50} = 4.97$ mg/L), 15 ($EC_{50} = 3.33$ g/L), and 16 ($EC_{50} = 3.43$ mg/L).

The above SARs of the 3-quinolinyl-4-chromenones 3-22 against S. sclerotiorum, V. mali, and B. cinerea could be

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ble 3. Fungicio	al Activities of th	e Target Con	npounds with	EC ₅₀ Value	s against S. sclero	tiorum, V. mali, and B.	cınerea
fungi	compd	R_1	R_2	EC ₅₀	95% CI ^a	regression equation	R^2
5. sclerotiorum	5	Н	5-F	4.60	2.34-4.87	y = -0.673 + 1.205x	0.99
	8	Н	5-OCH ₃	3.79	1.03-6.35	y = -0.908 + 1.571x	0.98
	9	Н	6-CH ₃	23.49	14.63-51.40	y = -2.528 + 1.844x	0.99
	10	Н	6-F	7.50	5.94-9.14	y = -1.237 + 1.414x	0.97
	13	Н	6-OCH ₃	3.65	2.74-4.51	y = -1.001 + 1.781x	0.98
	15	Н	7-OCH ₃	5.27	4.45-6.22	y = -1.066 + 1.477x	0.95
	16	Н	8-OCH ₃	8.55	6.55-10.75	y = -1.088 + 1.167x	0.99
	17	Н	3-OCH ₃	18.18	15.72-21.29	y = -2.507 + 1.989x	0.99
	21	CH ₃	6-OCH ₃	7.82	5.73-10.01	y = -0.946 + 1.059x	0.94
	23	4-CH ₃	Н	4.20	3.54-4.96	y = -0.928 + 1.489x	0.94
	24	4-F	Н	4.42	2.18-7.15	y = -0.892 + 1.382x	0.99
	25	4-Cl	Н	1.94	1.61-2.29	y = -0.487 + 1.692x	0.99
	26	4-OCH ₃	Н	2.17	1.72-2.64	y = -0.449 + 1.338x	0.98
	28	3-CH ₃	Н	4.37	3.58-5.28	y = -0.796 + 1.243x	0.97
	29	Ph	Н	3.56	2.35-4.74	y = -0.696 + 1.262x	0.99
	chlorothalonil			1.57	0.95-2.22	y = -0.241 + 0.954x	0.96
	boscalid			0.67	0.47-0.89	y = 0.183 + 1.042x	0.97
7. mali	4	Н	5-CH ₃	3.55	0.03-8.13	y = -0.184 + 0.334x	0.99
	6	Н	5-Cl	2.97	1.83-4.09	y = -0.583 + 1.231x	0.98
	7	Н	5-Br	5.79	4.10-7.50	y = -0.859 + 1.126x	0.97
	8	Н	5-OCH ₃	1.65	1.36-1.95	y = -0.386 + 1.769x	0.98
	9	Н	6-CH ₃	3.38	2.54-4.32	y = -0.493 + 0.934x	0.98
	10	Н	6-F	3.47	2.77-4.20	y = -0.809 + 1.498x	0.96
	13	Н	6-OCH ₃	2.61	2.16-3.10	y = -0.630 + 1.511x	0.97
	15	Н	7-OCH ₃	3.33	2.47-4.24	y = -0.610 + 1.167x	0.97
	16	Н	8-OCH ₃	3.43	2.72-4.21	y = -0.619 + 1.156x	0.95
	17	Н	3-OCH ₃	4.97	3.87-6.18	y = -0.810 + 1.163x	0.95
	21	CH ₃	6-OCH ₃	6.81	5.43-8.41	y = -0.971 + 1.165x	0.98
	23	4-CH ₃	н	3.52	2.83-4.23	y = -0.842 + 1.543x	0.98
	24	4-F	Н	3.18	2.48-3.92	y = -0.705 + 1.401x	0.95
	25	4-Cl	Н	1.56	1.23-1.89	y = -0.286 + 1.490x	0.96
	26	4-OCH ₂	Н	3.02	2.37-3.69	y = -0.711 + 1.481x	0.98
	27	4-CF ₂	н	3.57	2.52-4.58	v = -0.820 + 1.486x	0.99
	28	3-CH ₂	н	3.98	3.21-4.80	y = -0.883 + 1.471x	0.95
	chlorothalonil	0 0003		11.24	8.20-15.12	y = -0.941 + 0.895x	0.99
	tebuconazole			0.27	0.16-0.40	$y = 0.528 \pm 0.937x$	0.96
3. cinerea	3	н	н	8.20	6.69-9.84	y = -1.396 + 1.528x	0.97
	8	н	5-0CH	2.37	1.64-3.17	y = -0.558 + 1.493x	0.95
	9	н	6-CH	25.87	19.42-38.46	y = -1.388 + 0.982x	0.95
	10	н	6-F	4 72	3 59-5 83	y = -1.049 + 1.558x	0.98
	13	н	6-0CH	2.32	1 38-3 31	y = -0.311 + 0.852x	0.98
	15	н	7-0CH.	2.92	1.92-3.03	y = -0.619 + 1.575x	0.98
	16	н	8.0CH	6.54	5 53 - 7 58	y = -1.693 + 2.076x	0.90
	17	н	3-0CH	13 54	9.53-7.58	$y = -0.862 \pm 0.761x$	0.99
	21		5-0CH	13.34	2 70 5 82	y = -0.302 + 0.701x	0.97
	21		0-0СП ₃	4.78	3.70 - 3.82	y = -0.754 + 1.364m	0.90
	25	4-C11 ₃	11 U	3.37	2.80-4.38	y = -0.734 + 1.304x	0.93
	24	4-F	п	4.52	3.00-3.44	y = -0.933 + 1.423x	0.93
	25 24	4-01	п	1.54	1.12-1.98	y = -0.209 + 1.125x	0.98
	20	4-0CH ₃	п	/.05	0.24-9.13	y = -1.410 + 1.59/x	0.99
	2/	4-CF ₃	п	0.45	4.09-8.24	y = -0.928 + 1.14/x	0.98
	28	3-CH3	н	5.28	3.08-0.86	y = -0.829 + 1.14/x	0.97
	29	Pn	н	38.48	25.11-82.53	y = -1.100 + 0./32x	0.98
	32	н	6-CH3	4.90	3.50-6.27	y = -0.874 + 1.268x	0.98

	Table 3. Fungicidal Activities of th	e Target Compound	s with EC ₅₀ Values against S. sci	lerotiorum, V. mali, and B. cinerea"
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^aConfidence interval.

summarized as follows: (1) The compounds bearing methoxy groups in the quinoline moiety show better fungicidal activities

chlorothalonil

procymidone

boscalid

than that with other groups (H, CH_3 , F, Cl, Br, and OCF_3). (2) The compounds with a 5/6-methoxy group in the

y = -0.485 + 1.043x

y = -0.059 + 0.581x

y = 0.891 + 0.905x

2.92

1.26

0.10

1.61 - 4.22

0.61-2.04

0.07-0.14

0.986

0.967

0.970

quinoline part exhibit similar antifungal activities, which are relatively higher than that with other groups. (3) The compounds including 6-methoxy quinoline were chosen for further lead optimization due to the availability of the starting material and cost considerations. When R_2 is fixed as a 6-methoxy group, the compound 13 with $R_1 = H$ at the 4-position of the phenyl ring displays the best antifungal activity.

As for the 3-isoquinolinyl-4-chromenones series, we initially explored the effect of the substituents on the phenyl ring. The results indicated that introduction of a chlorine atom on the phenyl ring (compound **25**) was favorable for increased fungicidal activities (Figure 5). Also, compound **25** displayed the most potent fungicidal activities ($EC_{50} = 1.94$, 1.56, 1.54 mg/L) against *S. sclerotiorum*, *V. mali*, and *B. cinerea*, respectively.



Figure 5. Summarized SARs of the target compounds against S. sclerotiorum, V. mali, and B. cinerea.

In Vivo Fungicidal Activities. In vivo fungicidal activity of the target compounds against B. cinerea was carried out on *Cucumis sativus Linn.* leaves. The efficacy of the antifungal protection is shown in Figure 6A. Compounds 13 and 25 showed promising protective activities of 88.24 and 94.12%, respectively. Notably, compounds 13 and 25 showed better antifungal effects than chlorothalonil (76.47%) but slightly worse than boscalid (97.06%) at the concentration of 50 mg/L. (Table 4).

Table 4. In Vivo Activity of the Target Compounds 13 and 25 against B. cinerea on Cucumis sativus Linn. Leaves (50 mg/L)

	protective activity				
compd	diameter of lesions (cm)	control efficacy (%)			
13	0.7 ± 0.1	88.24%			
25	0.6 ± 0.1	94.12%			
chlorothalonil	0.9 ± 0.2	76.47%			
boscalid	0.5 ± 0.1	97.06%			
СК	2.2 ± 0.2				

Effects of Compounds 13 and 25 on Cell Membrane Permeability of *S. sclerotiorum, V. mali*, and *B. cinerea*. The change of relative conductivity indicates the variation of cell membrane permeability. According to literature,^{34,36} the cell membrane of fungi is an important target for fungicides, which can inhibit the synthesis of a phospholipid bilayer or protein and cause the leakage of internal electrolytes.^{37,38} As shown in Figure 6B, after treatment with compounds 13 and 25, the mycelial relative conductivities of *S. sclerotiorum, V. mali*, and *B. cinerea* increased with time, and the increase extent is much greater than that with the control, which indicates that



Figure 6. (A) In vivo fungicidal activity of the target compounds against *B. cinerea* was carried out on *Cucumis sativus Linn.* leaves. (B) Mycelial relative conductivity of *S. sclerotiorum*, *V. mali*, and *B. cinerea* in the presence or absence of compounds 13 and 25.

compounds 13 and 25 could enhance the mycelial cell membrane permeability of *S. sclerotiorum, V. mali,* and *B. cinerea.* Combined with the change of mycelial morphology after treatment with compounds 13 and 25, it can also be inferred that compounds 13 and 25 could manifest antifungal properties by destroying the cell membrane structure. The above reported findings revealed that the action target site of compounds 13 and 25 on *S. sclerotiorum, V. mali,* and *B. cinerea* could be the cell membrane.

Scanning Electron Microscopy (SEM) Analysis. SEM of *S. sclerotiorum* and *B. cinerea* revealed morphological changes in the fungal cell surface. As shown in Figure 7, the surfaces of *S.*



Figure 7. Surfaces of S. sclerotiorum (A, E) and B. cinerea (I, M) cells in the untreated groups were relatively smooth and regular, whereas when treated with compounds 13 (B, F, J, N), 25 (C, G, K, O), and Boscalid (D, H, L, P) at 5 mg/L there were significant shrinkage.

sclerotiorum and *B. cinerea* cells in the untreated groups (A, E, I, and M) were relatively smooth and regular, whereas when treated with compounds 13 (B, F, J, and N) and 25 (C, G, K, and O), and boscalid (D, H, L, and P) at 5 mg/L, there were significant shrinkage. Of note, for the *B. cinerea*, the extent of mycelial shrinkage with compound 25 (Figure 7, K and O) is greater than that with compound 13 (Figure 7, J and N). This showed that compounds 13 and 25 have a certain degree of damage to the cell wall of *B. cinerea*, and the damage intensity with compound 25 is greater.

Effects of Compounds 13 and 25 on the Respiratory Metabolism of *B. cinerea*. The respiratory inhibition effects of compounds 13 and 25, and boscalid on *B.cinerea* mycelia are shown in Table 5. Compounds 13 and 25 have inhibitory

Table 5. Respiratory Inhibition Rate of Compounds 13 and 25 and Boscalid on *B. cinerea* Mycelia

compd	R_0 (O ₂ , μ mol/g min)	R_1 (O ₂ , μ mol/g min)	IR (%)
13	3.83 ± 0.02	2.63 ± 0.03	32.16 ± 0.58
25	3.67 ± 0.03	2.87 ± 0.09	19.06 ± 1.43
boscalid	3.73 ± 0.03	2.53 ± 0.02	33.02 ± 0.59

effects on the respiration rate of *B. cinerea* mycelia, and their inhibition rates were 32.16 and 19.06%, respectively. The inhibition rate of compound **13** is close to that of boscalid (33.02%). We speculate that the antifungal mechanism of compound **13** may be similar to that of boscalid, which involves the inhibition of fungal respiration.

In summary, series of 3-(iso)quinolinyl-4-chromenone derivatives were designed and synthesized by the active substructure splicing principle and evaluated for their antifungal activities. The lead optimization was guided by bioactivity. The antifungal bioassay discovered the highly active compounds 13 and 25 with a broad spectrum of excellent in vitro fungicidal activities and better in vivo efficacy against B. cinerea. Physiological and biochemical studies showed that the primary action of mechanism of compounds 13 and 25 on S. sclerotiorum and B. cinerea may involve changing mycelial morphology and increasing cell membrane permeability. In addition, compound 13 may also affect the respiratory metabolism of B. cinerea. This study reveals that compounds 13 and 25 could be promising antifungal candidates and provide a valuable reference for further development of 3-(iso)quinolinyl-4-chromenones in crop protection.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jafc.0c06700.

The physical and spectroscopic data of 1a-h, 3-33 (PDF)

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Notes

The authors declare no competing financial interest.

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